## **Supporting Information**

## Density Functional and Spin-Orbit *Ab Initio* Study of CF<sub>3</sub>Br: Molecular Properties and Electronic Curve-Crossing

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	HF	MP2	CCSD(T)	B3LYP	B3PW91	PBE0	mPW1PW91	B97-2	ωB97X	B1B95	TPSSh	BMK	M05- 2X	M06- 2X
r(C-F) <sup>b</sup>	Х	0	0	0	0			Ο	0		0		0	
$\angle (F-C-F)^b$		0	0						0	0	0	Ο		0
Vib. Freq. <sup>b</sup>	Х				Ø			Ø	0			Х		Х
r(C–F) <sup>c</sup>	Х	0	0	0	0			0	0		0		0	
$r(C-Br)^{c}$	0	Х	Ο	Х					0	0	Х		0	
$\angle (F-C-F)^c$	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Vib. Freq. <sup>c</sup>	Х				ø	0	0	0		Ο				
$\mu^{c}$	Х	Х	Х		Х	Х	Х	Х	0	Х	Х	0	0	0
${D_0}^{\rm c}$	Х	Х	0	Х		0		0		Ο			0	

Table S1. The performance of methods for various molecular properties.<sup>a</sup>

a: (Best)  $@(4) \rightarrow O(3) \rightarrow \Box(2) \rightarrow X(1)$  (Worst)

b: molecular structure and property of CF<sub>3</sub> radical.

c: molecular structure and properties of  $CF_3Br$ .

Figure S1. The active orbitals of  $CF_3Br$  for CAS(6,6) calculations.



Figure S2. The scatter plot of the percentage error of the C–Br bond length (x-axis) vs. MAEs of  $T_v$  values (y-axis). The correlation coefficient is 0.909.

