

Exploring Fine Structures of Photoactive Yellow Protein in Solution Using Wide-Angle X-ray Scattering[†]

Tae Kyu Kim, Xiaobing Zuo,[‡] David M. Tiede,[‡] and Hyotcherl Ihee*

Department of Chemistry and School of Molecular Science (BK21), KAIST, Daejeon 305-701, Korea

[‡]Chemistry Division, Argonne National Laboratory, Argonne, Illinois 60439, USA

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We demonstrate that wide-angle X-ray scattering pattern from photoactive yellow protein (PYP) in solution using a high flux third generation synchrotron X-ray source reflects not only the overall structure, but also fine structures of the protein. X-ray scattering data from PYP in solution have been collected in q ranges from 0.02 \AA^{-1} to 2.8 \AA^{-1} . These data are sensitive to the protein structure and consistent with the calculation based on known crystallographic atomic coordinates. Theoretical scattering patterns were also calculated for the intermediates during the photocycle of PYP to estimate the feasibility of time-resolved wide-angle X-ray scattering experiments on such proteins. These results demonstrate the possibility of using the wide-angle solution X-ray scattering as a quantitative monitor of photo-induced structural changes in PYP.

Key Words : Wide-angle X-ray scattering, Photoactive yellow protein, Time-resolved, Biological macromolecule, 3-Dimensional structure

Introduction

The major purposes of structural molecular biology include identifying structural changes of biological macromolecules and correlating these changes to their biological functions. Over the past years, three-dimensional structures of vast number of biological molecules have been determined using X-ray crystallography and nuclear magnetic resonance (NMR).¹ However, these high resolution methods have their own limitations and therefore can be applied only when rather specific conditions are met. For example, structural determination by X-ray crystallography requires high-quality protein crystals whose preparation sets one of the major bottlenecks. NMR overrides this requirement and allows structures in solution to be studied, but the size of the protein typically accessible by NMR is still much smaller than that of X-ray crystallography.

In addition, the majority of the determined 3-D structures represent the ground state of bio-molecules. While these static structures carry enormous amount of information, the structures of intermediates should be known in order to correlate the structures to their functions and to understand the detailed mechanisms.^{2,3} To this goal, X-ray crystallography has progressed to adopt time-resolved studies, and recently picosecond time-resolution using this technique has been demonstrated.⁴⁻⁷ With this success, time-resolved X-ray crystallography holds a great promise in its applications to determine the detailed 3-D structures of transient intermediates. At this point, one important question arises among scientists in this field: whether the intermediate structures determined by time-resolved X-ray crystallog-

raphy correctly represent the reality in solution. To answer this question and to provide a complementary tool for time-resolved studies of macromolecules, we propose to utilize wide-angle X-ray scattering in solution.

Solution X-ray scattering of protein⁸ has been widely used to determine overall structure of native particle in various physiological conditions and conformational change due to the variation of external parameters. Usually small-angle scattering region is used to determine general structural features such as the overall size and shape. This small-angle X-ray scattering (SAXS) technique^{9,10} has achieved considerable success in restoring three dimensional structures of proteins from the scattering patterns. However, the limited information in the relatively small q region, *i.e.* $q \leq 0.2 \text{\AA}^{-1}$, (where $q = 4\pi\sin\theta/\lambda$ is the magnitude of scattering vector, θ is half of the scattering angle, and λ is the wavelength) is an obstacle to the capability to restore three dimensional structures.¹¹ Therefore the measurement of scattering profile for a variety of proteins should be extended to high q region with high accuracy. It is important to detect wide-angle (high-angle and wide-angle are used interchangeably in this article) scattering patterns since it directly probes distance correlations on length scales that are small compared to the dimensions and contains rich information of detailed fine structure of protein in solution. The practical use of high q region data in the solution scattering experiments is limited by the difficulty of measuring the weak protein scattering signal which is averaged out with a much stronger background signals from the solvents and solution container. However, the recent advancements of high-flux third generation synchrotron X-ray source and detectors make the measurement of wide-angle scattering patterns for proteins in solution experimentally possible.^{12,13} Recently, it was reported not only that wide-angle scattering patterns from insertion-device beamline at third generation X-ray synchro-

[†]Dedicated to Professor Yong Hae Kim for his distinguished achievements in organic chemistry.

*To whom correspondence should be addressed. e-mail: Hyotcherl.Ihee@kaist.ac.kr

tron sources are sensitive to protein conformation states,¹⁴ but also that the measured scattering patterns can be compared quantitatively with calculated patterns from detailed structural models.^{14,15}

If the time-resolved X-ray wide angle scattering is realized in solution, it will have many advantages over the time-resolved X-ray crystallography. First, it does not require highly-diffracting crystals and this will widen its applicability to a variety of systems. Second, any desired conditions including near-physiological one can be easily realized. Third, irreversible reactions, which are hard to study with the current time-resolved crystallography, can be studied since a simple flow cell will provide a fresh sample for each laser shot or other perturbation.

Photoactive yellow protein (PYP)¹⁶ is a small (125-residue, 14 kD) blue light-sensitive protein found in the purple phototrophic bacterium *Halorhodospira halophila*. It has been proposed as a blue light receptor for negative phototaxis of the bacterium based on the similarity of the visible absorption profile of the protein and the action spectrum of the organism.¹⁷ Since PYP is the first example of photoreceptor protein responsible for phototropic signal transduction, it is one of the most suitable targets for studying the light-capturing mechanism of living cells at atomic level. For this reason, PYP has been extensively investigated by spectroscopy¹⁸ and X-ray crystallography.^{19,21} Upon excitation by blue light, the chromophore of PYP, *trans* p-coumaric acid, switches to the *cis* form.²² Eventually this returns to the original form, completing a photocycle (Figure 1).^{23,24} The phototropic signal transduction pathway is not known, but the conformational change occurring on the formation of the pB state and its relaxation to the dark state pG are suggested to be responsible for relaying the signal to the next step of activation mechanism.²⁵ The

ground state structure of PYP has been determined by X-ray crystallography²⁰ and by solution NMR.²¹ However, there are noticeable differences between crystal structure of the pG state of PYP determined by X-ray crystallography and the ensemble of solution structures revealed by the NMR spectroscopy. In addition, X-ray crystallographic structures are also available for the stationary pB state.²⁶

In this work, we have carried out wide-angle solution scattering experiments of PYP, to elucidate an availability of the wide-angle synchrotron radiation X-ray scattering. Here we demonstrate that X-ray scattering data from proteins in solution covering the q range from 0.02 \AA^{-1} to 2.8 \AA^{-1} are consistent with the calculation¹⁵ based on known crystallographic atomic coordinates and NMR structures. We show that scattering patterns in high angle region are very sensitive to fine structures of a protein in solution. We also report simulated scattering patterns from the pG and pB states of PYP for time-resolved solution X-ray scattering.

Material and Method

Sample Preparation. Apo-PYP was expressed^{19,22} in *Escherichia coli* and reconstituted²⁷ with the chromophore to yield holo-PYP as described in the literatures. Finally PYPs were suspended in 50 mM phosphate (pH 7.0) and 20 mM NaCl buffer at the concentration of 30 mg/mL.

X-ray Scattering Measurements. Wide angle scattering experiments were carried out at the undulator beam line 12-ID at the Advanced Photon Source (APS), Argonne National Laboratory. The X-ray scattering instrument utilized a double-crystal Si(111) monochromator and a two-dimensional mosaic CCD detector for detection.¹² The X-ray wavelength was set at $\lambda = 0.685 \text{ \AA}$ and the sample-to-detector distance was adjusted to measure scattering across the range of momentum transfer $0.02 \leq q \leq 2.8 \text{ \AA}^{-1}$. Protein solution and buffer background two-dimensional scattering patterns were measured as the average of 10 X-ray scattering images collected at 1 sec intervals with 0.1 sec exposure times using a 1.5 mm capillary flow cell and a sample flow rate of $5 \times 10^{-5} \text{ l/min}$. One-dimensional, azimuthally-averaged $I(q)$ scattering patterns were determined from the scattering images, and protein scattering patterns were obtained by the X-ray transmission normalized subtraction of the buffer scattering pattern. Details of the 12ID beamline,²⁸ X-ray scattering instrument,¹² and protein X-ray scattering measurements²⁸ have been previously described.

Calculation of Scattering Patterns. The general theory of solution X-ray scattering can be found in many literatures^{9,29} and a brief summary is given in this section. Solution scattering pattern, $I(q)$, can be represented as follows:

$$I(q) = nP(q)S(q) \quad (1)$$

where n is the concentration of macromolecule in solution, $P(q)$ is the macromolecular form factor, and $S(q)$ is the solution structure factor. Here, we consider protein in dilute solution. Under this condition, we can assume that $S(q) \approx 1$ and the scattering patterns can be interpreted as form factors

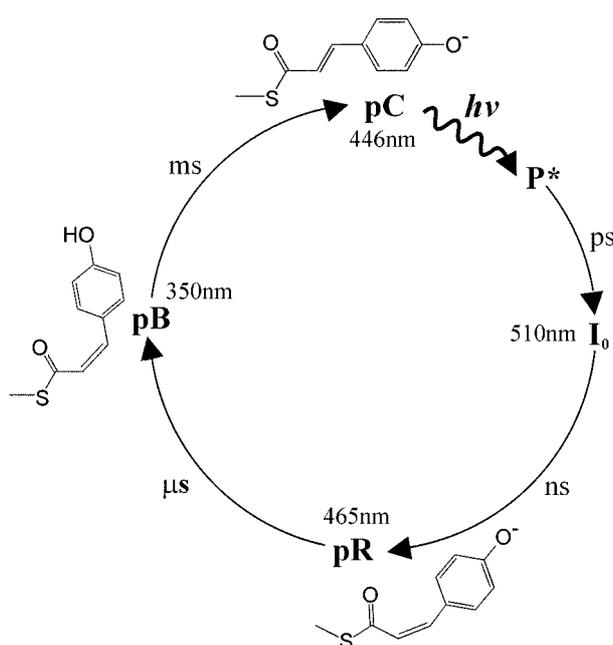


Figure 1. Photocycle of PYP in solution. pG: ground state; pR: red-shifted state; pB: blue-shifted signal transduction state.

calculated from atomic coordinates. The method of calculating scattering patterns using atomic coordinates was first developed by Svergun *et al.*,⁹ and was extended to produce accurate calculation of scattering patterns in high angle region using numerical precision.^{14,15} In this approach, protein scattering is calculated from average of partial scattering amplitudes from the fixed atomic coordinated in the molecules, A_a , and the solvent excluded volume, A_s

$$P(q) = \langle |A_a(q) - A_s(q)|^2 \rangle_{\Omega} \quad (2)$$

where Ω represents the unit solid angle. The coherent scattering amplitude for the protein including the Debye-Waller atomic fluctuation factor²⁹ is given by:

$$A_a(q) = \rho_0 \sum_{j=1}^N f_j(q) e^{iq \cdot r_j} e^{-B_j q^2 / 16\pi^2} \quad (3)$$

where f_j is the atomic form factor, B_j is the Debye-Waller B-factor, and N is the total number of non-hydrogen atoms. Hydrogen atoms, which make a significant contribution to the total protein scattering are grouped to the corresponding non-hydrogen atoms. f_j for the group is calculated from the group atomic geometry. The Debye-Waller B-factor is proportional to the mean square position fluctuation and to the absolute temperature.

The solvent excluded volume was estimated by the dummy atom model.^{9,30} The total scattering amplitude for the dummy atom is:

$$A_s(q) = \rho_0 \sum_{j=1}^N g_j(q) e^{iq \cdot r_j} \quad (4)$$

where ρ_0 is solvent electron density and g_j is the form factor for the dummy atoms with volume V_j :

$$g_j(q) = G(q) V_j e^{-q^2 V_j^{2/3} / 4\pi} \quad (5)$$

and $G(q)$ is a volume expansion factor⁹ given as:

$$G(q) = \frac{V_0}{V_m} e^{-q^2 (V_0^{2/3} - V_m^{2/3}) / 4\pi} \quad (6)$$

where V_0 is the expanded atomic volume and V_m is the average atomic volume for the group.

Using these calculation procedures, we demonstrate the sensitivity of wide-angle scattering patterns to the details of protein structures.

Results and Discussion

Static Wide-angle Scattering Pattern on PYP in Solution. Figure 2 shows experimental and calculated scattering patterns for PYP in solution. The observed scattering pattern in the small angle region below $q \leq 0.11 \text{ \AA}^{-1}$ shows a broad peak due to the electrostatic based interparticle interaction. For the globular particle systems in solution, the repulsive interparticle interaction usually produces a broad single peak in the small q region.³¹ The

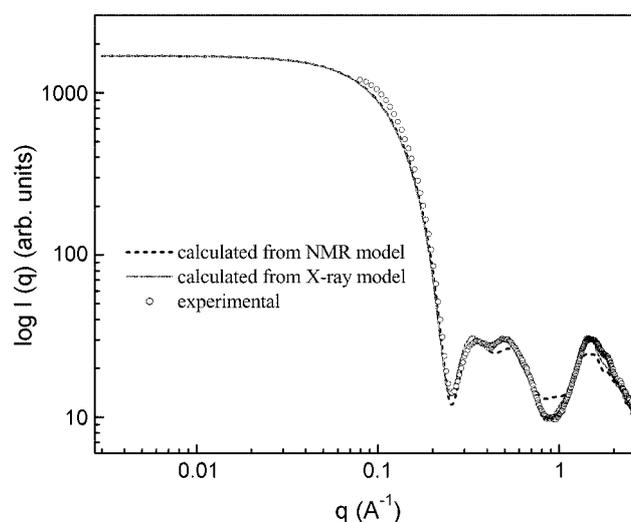


Figure 2. Comparison of experimental and calculated scattering patterns from X-ray and NMR model. Experimental scattering patterns were measured with protein concentration of 30 mg/mL for PYP.

amplitude and position of the broad shoulder peak depend on nonrandom particle distribution in solution, characterized by the solution structure factor. In high ionic strength buffers, electrostatic screening reduces the interaction, and therefore scattering profile for protein accurately reflects the protein form factor, not the solution structure factor (*i.e.*, $I(q) = nP(q)S(q)$, where $S(q) \rightarrow 1$). The experimental scattering pattern for PYP is in good agreement with the calculated scattering pattern based on the crystal atomic coordinate in overall q range. A slight deviation in small angle region reflects interparticle interaction have not been minimized in our experiment.^{14,32} However, the scattering pattern in high angle region is not altered by interparticle interaction.¹⁴

There are two different structural models for PYP determined by X-ray crystallography³³ and solution NMR spectroscopy.²¹ The calculated scattering profile from the NMR model is the average of 26 similar structures. There is no Debye-Waller factor for the NMR structure in the pdb file, therefore the calculated scattering patterns from the NMR and X-ray crystallographic model were obtained without Debye-Waller effects for comparison. The scattering patterns for structures determined by these two techniques show significant differences at high angle regions, but they are indistinguishable at small angle, $q \leq 0.2 \text{ \AA}^{-1}$. This finding can be explained by the fact that the X-ray and NMR models have similar overall size and shape, but have different fine structures. This implies that scattering pattern in high angle region is very important in depicting fine structures of protein and structural dynamics. A major factor affecting the minor discrepancy between experimental scattering patterns and those calculated from the NMR model is the unweighted averaging of structures across the NMR ensembles. We note that scattering patterns calculated for the individual structures within this ensemble show a wide variation, with only a subset of the structures being compatible with experimental data.

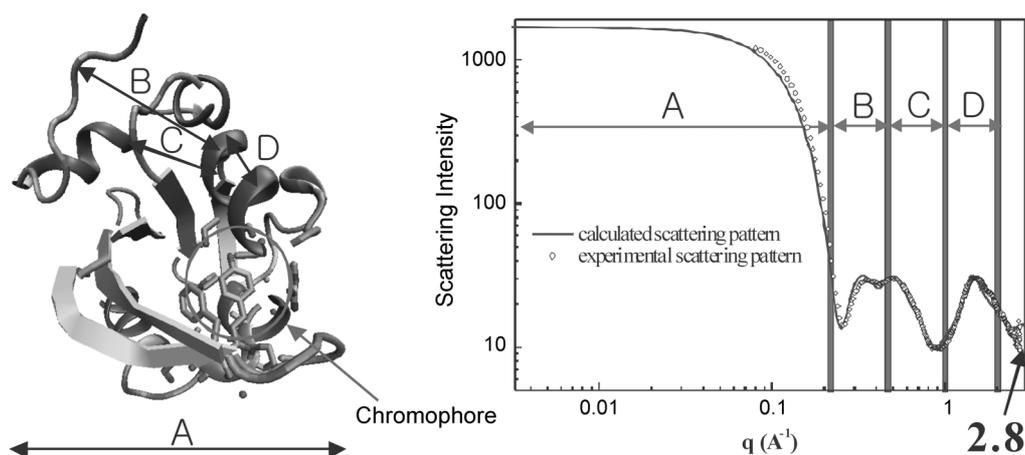


Figure 3. The regions of the experimental and calculated X-ray scattering patterns on PYP which corresponds to typical intramolecular distance in structure of PYP are shown.

The wide-angle scattering pattern for PYP in solution shows noticeable oscillatory feature which is in good agreement with the calculated scattering pattern from crystal atomic coordinates. The minor disagreement at high angle region can be attributed to the fact that PYP in solution may be composed of many degrees of ensemble structures rather than one fixed and rigid structure.²¹ The thermal motion of a protein trapped in the crystal lattice would be more restricted than that in solution. Therefore, thermal motions of proteins in solution can be enhanced than those in crystal, which contribute more efficiently to the smearing of the experimental scattering patterns at high-angle region and this effect should be taken into account.^{14,15,34} In the simulation, the thermal effect is considered using crystallographic B-factor multiplied by a scaling factor to match the calculated and experimental high angle scattering patterns. This correction accounts for the effect due to individual incoherent atomic thermal factors. Other types of motions anticipated for proteins, including the rigid-body movement of groups of atoms, can significantly alter high-angle scattering patterns.^{14,34,35} It was reported that the calculated scattering pattern for horse cytochrome c required the B factor multiplied by a factor 3 to match the experimental scattering factor,¹⁴ which indicates the existence of significant configurational fluctuations for this protein in solution. However, in the case PYP, the best fit was obtained without using any B-factors, which suggests that the PYP in its pG state may be thermally very stable and compact compared with other proteins. The temperature factor affects significantly on the scattering at high angle region and has negligible effects on small angle scattering pattern, implying that scattering profiles in wide-angle region are well indicative of dynamic structure of protein in solution. Therefore, wide-angle scattering patterns can be explained with the calculated patterns using atomic crystallographic coordinate provided that the thermal effect is properly considered.^{14,34}

Comparison of scattering patterns of other proteins¹³ in solution with that of PYP shows that there is significant difference of shape in the q range from $\sim 0.2 \text{ \AA}^{-1}$ to $\sim 2.8 \text{ \AA}^{-1}$.

This suggests that wide-angle scattering patterns reflect detailed intramolecular structures of proteins and can serve as a finger print region. The scattering patterns in the q range from $\sim 0.2 \text{ \AA}^{-1}$ to $\sim 0.7 \text{ \AA}^{-1}$ and the range from $\sim 0.7 \text{ \AA}^{-1}$ to $\sim 2.8 \text{ \AA}^{-1}$ correspond to the domain structure and the secondary structure, respectively. In Figure 3, domains in the

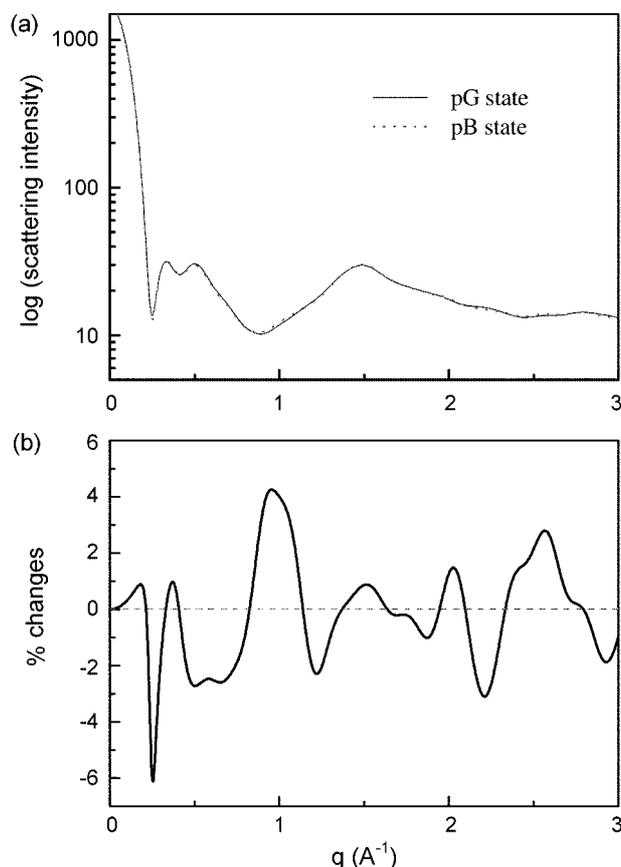


Figure 4. (a) Comparison of calculated scattering patterns for pG and pB states of PYP. Scattering patterns for pG and pB states are shown in solid and dotted line, respectively. (b) pB-minus-pG difference scattering patterns calculated from crystallographic atomic coordinates of PYP.

scattering pattern are roughly assigned according to intramolecular distances of PYP. However, one should aware that this classification is not definite and in principle, all inter-nuclear distances contribute to the overall scattering angles.

Calculation of Light-induced Changes in Wide Angle Scattering. The light-induced change in PYP provides a useful target to probe the sensitivity of wide-angle scattering to the fine structural changes of the protein during its photocycle. Scattering patterns on pG and pB states of PYP based on structural models from crystallographic study³³ were calculated and shown in Figure 4(a). This result suggests that wide-angle scattering patterns should be sensitive to structural difference between pG and pB states if they are measured with sufficient precision and resolution. The difference of scattering patterns is shown in Figure 4(b), plotted as the percentage change in scattered intensity. The difference of scattering pattern calculated from the pG and pB states of PYP coordinates shows an oscillating structure with maximum amplitude to about 5% change in scattered intensity. The high X-ray flux from the third generation synchrotron source permits high precision measurements. For example, the average standard deviation measured with respect to the amplitude of scattering in the region $0.2 < q < 1.0 \text{ \AA}^{-1}$ was 0.4%. Therefore, experimental difference scattering patterns in high-angle region can be measured with the sufficient sensitivity to elucidate the state-dependent small structural changes of PYP. Work is under progress to measure the time-resolved wide-angle X-ray scattering patterns of PYP using a third generation synchrotron source.

Conclusion

The static wide-angle X-ray scattering pattern of PYP in solution in the range from $\sim 0.08 \text{ \AA}^{-1}$ to $\sim 2.8 \text{ \AA}^{-1}$ was measured using a third generation synchrotron X-ray source. Wide-angle scattering patterns are sensitive to the detailed structures of protein such as domain structures, secondary structures, and their packing. Furthermore these wide-angle scattering patterns give a chance to make a comparison between experimental scattering patterns and detailed structural models determined by the X-ray crystallography. X-ray scattering profiles for the intermediate (pB state) and ground state (pG) of PYP were calculated. Wide-angle X-ray scattering obtained with the high precision should be capable of monitoring light-induced changes in conformation and structural dynamics of PYP.

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References

1. Edwards, A. M.; Arrowsmith, C. H.; Christendat, D.; Dharamsi, A.; Friesen, J. D.; Greenblatt, J. F.; Vedadi, M. *Nat. Struct. Biol.* **2000**, *7*, 970.
2. Kim, S.; Jin, G.; Lim, M. *Bull. Korean Chem. Soc.* **2003**, *24*, 1470.
3. Lim, M. *Bull. Korean Chem. Soc.* **2002**, *23*, 865.
4. Wulff, M.; Plech, A.; Eybert, L.; Randler, R.; Schotte, F.; Anfinrud, P. *Faraday Discussions* **2003**, *122*, 13.
5. Schotte, F.; Lim, M. H.; Jackson, T. A.; Smirnov, A. V.; Soman, J.; Olson, J. S.; Phillips, G. N.; Wulff, M.; Anfinrud, P. A. *Science* **2003**, *300*, 1944.
6. Techert, S.; Schotte, F.; Wulff, M. *Phys. Rev. Lett.* **2001**, *86*, 2030.
7. Bratos, S.; Mirloup, F.; Vuilleumier, R.; Wulff, M. *J. Chem. Phys.* **2002**, *116*, 10615.
8. Guinier, A. Fournet, G. *Small-Angle Scattering of X-rays*; John Wiley & Sons; New York, 1995.
9. Svergun, D.; Barberato, C.; Koch, M. H. J. *J. Appl. Cryst.* **1995**, *28*, 768.
10. Chacon, P.; Moran, F.; Diaz, J. F.; Pantos, E.; Andreu, J. M. *Biophys. J.* **1998**, *74*, 2760.
11. Svergun, D. *Biophys. J.* **1999**, *76*, 2879.
12. Fischetti, R. F.; Rodi, D. J.; Mirza, A.; Irving, T. C.; Kondrashkina, E.; Makowski, L. *J. Synchrotron Rad.* **2003**, *10*, 398.
13. Hirai, M.; Iwase, H.; Hayakawa, T.; Miura, K.; Inoue, K. *J. Synchrotron Rad.* **2002**, *9*, 202.
14. Tiede, D. M.; Zhang, R.; Seifert, S. *Biochemistry* **2002**, *41*, 6605.
15. Zhang, R.; Thiyagarajan, P.; Tiede, D. M. *J. Appl. Cryst.* **2000**, *33*, 565.
16. Meyer, T. E. *Biochem. Biophys. Acta* **1985**, *806*, 175.
17. Sprenger, W. W.; Hoff, W. D.; Armitage, J. P.; Hellingwerf, K. J. *J. Bacteriol.* **1993**, *175*, 3096.
18. Imamoto, Y.; Mihara, K.; Tokunaga, F.; Kataoka, M. *Biochemistry* **2001**, *40*, 14336.
19. Baca, M.; Borgstahl, G. E. O.; Boissinot, M.; Burke, P. M.; Williams, D. R.; Slater, K. A.; Getzoff, E. D. *Biochemistry* **1994**, *33*, 14369.
20. Borgstahl, G. E. O.; Williams, D. R.; Getzoff, E. D. *Biochemistry* **1995**, *34*, 6278.
21. Dux, P.; Rubinstenn, G.; Vuister, G. W.; Boelens, R.; Mulder, F. A.; Hard, K.; Hoff, W. D.; Kroon, A. R.; Crielaard, W.; Hellingwerf, K. J.; Kaptein, R. *Biochemistry* **1998**, *37*, 12689.
22. Kort, M.; Vonk, H.; Xu, X.; Hoff, W. D.; Crielaard, W.; Hellingwerf, K. J. *FEBS Lett* **1996**, *382*, 73.
23. Imamoto, Y.; Kataoka, M.; Tokunaga, F.; Asahi, T.; Masuhara, H. *Biochemistry* **2001**, *40*, 6047.
24. Ujj, L.; Devanathan, S.; Meyer, T. E.; Cusanovich, M. A.; Tollin, G.; Atkinson, G. H. *Biophys. J.* **1998**, *75*, 406.
25. Hoff, W. D.; Xie, A.; Van Stokkum, I. H. M.; Tang, X.; Gural, J.; Kroon, A. R.; Hellingwerf, K. J. *Biochemistry* **1999**, *38*, 1009.
26. Genick, U. K.; Soltis, S. M.; Kuhn, P.; Canestrelli, I. L.; Getzoff, E. D. *Nature* **1998**, *392*, 206.
27. Imamoto, Y.; Ito, T.; Kataoka, M.; Tokunaga, F. *FEBS Lett.* **1995**, *374*, 157.
28. Beno, M. A.; Jennings, G.; Engbretson, M.; Knapp, G. S.; Kurtz, C.; Zabransky, B.; Linton, J.; Seifert, S.; Wiley, C.; Montano, P. A. *Nucl. Instrum. Meth. A* **2001**, *467*, 690.
29. Warren, B. E. *X-ray Diffraction*; Dover Publication, Inc.: New York, 1990.
30. Fraser, R. D. B.; MacRae, T. P.; Suzuki, E. *J. Appl. Cryst.* **1978**, *11*, 693.
31. Hirai, M.; Iwase, H.; Hayakawa, T. *J. Phys. Chem. B* **1999**, *103*, 10136.
32. Ducruix, A.; Guilloteau, J. P.; Ries-Kautt, M.; Tardieu, A. *J. Cryst. Growth* **1996**, *168*, 28.
33. Genick, U. K.; Borgstahl, G. E.; Ng, K.; Ren, Z.; Pradervand, C.; Burke, P. M.; Srajer, V.; Teng, T. Y.; Schildkamp, W.; McRee, D. E.; Moffat, K.; Getzoff, E. D. *Science* **1997**, *275*, 1471.
34. Zhang, R.; Marone, P. A.; Thiyagarajan, P.; Tiede, D. M. *Langmuir* **1999**, *15*, 7510.
35. Tiede, D. M.; Zhang, R.; Chen, L. X.; Yu, L.; Lindsey, J. S. *J. Am. Chem. Soc.* **2004**, *126*, 14054.